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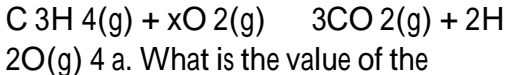
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Answer the following questions in the space provided. 1. Given the following equation:



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coefficient x in this equation? 40.07 g/mol b.
What is the molar mass of C₃H₄? 2 mol O
2:1 mol H₂O c. What is the mole ratio ...

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Stoichiometry Answers Mix & match (both
simple stoichiometry and limiting reagent

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problems) Units to use (select at least one):
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formulas or names: Formulas only Names
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Write the answer on the line to the left.

Show all your work in the space provided. 1.

_____ The actual yield of a reaction is 22 g
and the theoretical yield is 25 g. Calculate

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the percentage yield. 2. 6.0 mol

Answers

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deals with the mass relationships of elements in compounds. reaction stoichiometry involves the mass relationships between reactants and products in a chemical reaction.

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HOMEWORK 9-2 (pp. 280-282)

VOCABULARY Complete each sentence.

1. ... the provided a 1 MODERN

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