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Worksheet ANSWER KEY. Write the
empirical formula for the following

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Compounds. 1) C_6H_6 CH_6 6) C_8H_{18}
 C_4H_9 7) WO_2 WO_2 8) $C_2H_6O_2$ CH_3O 9) X_3Y_{13} X_3Y_6 6) A compound with an empirical formula of C_2OH_4 and a molar mass of 88 grams per mole. What is the molecular formula of this compound? $C_4O_2H_8$ 7) A compound with an empirical formula of

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C. 4. H. 4. O and a molar mass of 136 grams per mole.

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October 23, 2020. Some of the worksheets below are Empirical Formula Worksheets, several exam style questions like determine the empirical formula of a compound that

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contains 53.70% iron and 46.30% sulfur, 3 solutions are provided at the end of each worksheet.

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Worksheet #8 Empirical Formulas 1.

State the empirical formula for each of
the following compounds: a) C_4H_8 ; b)
 $C_2H_6O_2$; c) N_2O_5 ; d) $Ba_3(PO_4)_2$; e)

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Worksheet #8 Empirical Formulas
2. What is the empirical formula for a compound that contains 0.063 mol chlorine and 0.22 mol

Worksheet #8 Empirical Formulas H O
N O 4I

EMPIRICAL AND MOLECULAR
FORMULA WORKSHEET An oxide of

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Worksheet Answers
Chromium is found to have the following % composition: 68.4 % Cr and 31.6 % O. Determine this compound's empirical formula. The percent composition of a compound was found to be 63.5 % silver, 8.2 % nitrogen, and 28.3 % oxygen. Determine the compound's empirical

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Molecular Formula. Empirical Formula.
Formaldehyde. CH_2O . CH_2O .
Acetic Acid. $\text{C}_2\text{H}_4\text{O}_2$. CH_2O .
Glucose. $\text{C}_6\text{H}_{12}\text{O}_6$. CH_2O

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to understand how to project cash flow. Regardless of what your company planning goals, cash flow is the most crucial resource in the company, and money is the one most important small business purpose.

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The empirical formula of a compound
is CH_2 , Its molecular mass is 70

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g/mol. What is its molecular formula?
= XCl) 70 C H A compound is found to
be 40.00% carbon, 6.7% hydrogen
and 53.5% oxygen. Its molecular mass
is 60. g/mol, What is its molecular
formula? 401) c 3133 4 A compound is
64.9% carbon, 13.5% hydrogen and
21.60% oxygen. Its molecular

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DETERMINING EMPIRICAL

FORMULAS Name What is the
empirical ...

The empirical formula of benzene is
CH (its molecular formula is C_6H_6).

If 10.00 mg of benzene is subjected to
combustion analysis, what mass of CO

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Worksheet Answers
2 and H₂O will be produced?

Answer: The empirical formula is C₄H₅. (The molecular formula of xylene is actually C₈H₁₀.) 33.81 mg of CO₂; 6.92 mg of H₂O

Chapter 11.2: Empirical and Molecular
Formulas - Chemistry ...

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Worksheet Answers, what is the empirical and molecular formula of serine? C = 34.95 % H= 6.844 % O = 46.56 % N= 13.59 % Percent Composition and Molecular Formula Worksheet Key

Percent Composition and Molecular

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Worksheet Answers

Molecular formula is 2 times the empirical formula $2(C_2O_4H_8) = C_4O_8H_{16}$

7) A compound with an empirical formula of C_4H_4O and a molar mass of 136 grams per mole. What is the molecular formula of this compound?

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Empirical and Molecular Formula
Worksheet

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Worksheet ANSWER KEY. Write the
empirical formula for the following
compounds. 1) C_6H_6 2) CH_4 3) C_8H_{18}
4) C_4H_9 5) WO_2 6) WO_2 7) $C_2H_6O_2$ 8) CH_4

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3 O 9) X39Y13 X 3 Y 6) A compound with an empirical formula of C. 2. OH. 4. and a molar mass of 88 grams per mole. Page 1/3

Determining Molecular Formulas
Answer Key

The empirical formula of a compound

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Worksheet Answers
represents the simplest whole-number ratio between the elements that make up the compound. This 10-question practice test deals with finding empirical formulas of chemical compounds. A periodic table will be required to complete this practice test. Answers for the test appear after the

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