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date:\_\_\_\_\_. CHAPTER 4 TEST: Atoms, Atomic Theory and Atomic Structure. Matching. A. Bohr B. Democritus C. Rutherford D. Dalton E. Thomson F. Schrodinger. \_\_\_\_\_ 1. Greek thinker; called nature 's basic particle an atom, based on the Greek word " atomos " which means " indivisible " . Did not have evidence that atoms existed.

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Questions : 1. The number of electrons in  $\text{Na}^+$  (atomic number = 11) are (a) 11 (b) 10 (c) 12 (d) 13 2. There are four elements P, Q, R and S having atomic numbers of 4, 18, 10 and 16 respectively. The element which can exhibit covalency as well as electrovalency will be: (a)...

Chapter 4 Structure of Atom MCQ Test 2 Science| Class 9th ...

4.05 Atomic Number, Mass number, Isotopes and Isobars. Chapter 4 Structure of Atoms - Test. 05 The Fundamental Unit of Life 5 Topics | 6 Quizzes 5.01 Introduction to cell 5.01 Introduction to Cell. 10.03 Plasma Membrane 5.02 Plasma Membrane. 10.04 Cell wall and Nucleus

Atomic Structure | Class 9 Chapter 4 Structure of Atom ...

The four atomic species can be represented as follows. Out of these, the two species which can be termed isobars are: (i)  $201 \text{ X } 60$  (ii)  $200 \text{ X } 61$  (iii)  $200 \text{ X } 58$  (iv)  $203 \text{ X } 60$

Chapter 4 Structure of Atom MCQ Test 1 Science| Class 9th ...

Atomic Structure (Chapter Test A) STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. emily\_reidt. Terms in this set (21) the total number of protons and neutrons in the nucleus of an atom. mass number. the weighted average mass of the atoms in a naturally occurring sample of an element.

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Chemistry Chapter 4 Atomic Structure Test

Weighted Atomic mass is  $26.4 + 8.9 = \underline{\hspace{2cm}}$  amu. Title: Notes Chapter 4: The Structure of the Atom Author: Administrator Last modified by: Kelli Pearce Created Date: 9/15/2015 10:36:00 PM Company: MISD Other titles: Notes Chapter 4: The Structure of the Atom ...

Notes Chapter 4: The Structure of the Atom

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