

## Chapter 17 Study Guide Acids Bases

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### Chapter 17 Chemistry of Acids and Bases Study Guide

Chapter 17. Name \_\_\_\_\_ STUDY GUIDE. Acids and Bases Use with Text Pages 466-473. Use the words in the box to fill in the blanks in the paragraphs below. An acid is a substance that produces \_\_\_\_\_ in solution. Acids taste \_\_\_\_\_. Some acids can produce painful burns and damage tissues.

### Chapter 17

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Acids are named by changing the ending of the alkane to -oic acid. i.e. a 4 carbon acid goes from butane to butanoic acid Acid salts are named by changing the acid ending to -oate i.e the 4 carbon acid goes from butanoic acid to sodium butanoate Esters are named by changing the ending of the alkane to -oate and by using alkyl name i.e. the 4 carbon ester below is methyl butanoate(methyl for OCH 3 portion) O O Anhydrides are dehydrated esters that replace -oic acid with word anhydride Acyl ...

### Chapter\_17\_Study\_Guide\_1 Chapter 18 Study Guide 1 ...

Read Book Chapter 17 Study Guide Acids Bases Chapter 17 Study Guide Acids Chapter 17: Acids and Bases. acid. base. alkaloids. Arrhenius. sour taste, dissolves metals, turns blue litmus paper red, mor.... bitter taste, slippery feel, turns red litmus paper blue, more.... organic bases found in plants that are often poisonous.

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Carboxyl, amine, hydrogen and variable side group Describe two forces that help shape an amino acid chain? Peptide bond holds chains of amino acids together by linking amine and carboxyl groups; hydrogen bonds help fold proteins into shapes and hold them together. Shape depends on amino acids, polarity and R groups.

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•The most common strong acids are HCl, HBr, HI, HNO 3, HClO 4, and H 2SO 4. •Strong acids are strong electrolytes. •All strong acids ionize completely in solution. •Example: Nitric acid ionizes completely in aqueous solution. HNO 3(aq) + H 2O(l) H 3O +(aq) + NO 3 -(aq) •Since H+ and H 3O + are used interchangeably, we write HNO

### AP Chemistry – CHAPTER 16 STUDY GUIDE Acid-Base Equilibrium

There are 20 amino acids, but only 4 nucleotide bases in DNA. That means it takes at least 3 nucleotides to code for one amino acid, hence a codon (three amino acids). □ The code is read as an RNA code, 5' to 3'.

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